POZNAN UNIVERSITY OF TECHNOLOGY



EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS) pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

COURSE DESCRIPTION CARD - SYLLABUS

Course name		
Kinetics of liquids and gases		
Course		
Field of study	Year/Semester	
Mechanical and Automotive Engin	1/1	
Area of study (specialization)	Profile of study	
		general academic
Level of study		Course offered in
First-cycle studies	polish	
Form of study	Requirements	
part-time		elective
Number of hours		
Lecture	Laboratory classes	Other (e.g. online)
18		
Tutorials	Projects/seminars	
9		
Number of credit points		
3		
Lecturers		
Responsible for the course/lecturer: Respons		Responsible for the course/lecturer:
dr Edyta Janeba-Bartoszewicz		
email: edyta.janeba-		
bartoszewicz@put.poznan.pl		
tel. 616652497		
Faculty of Civil Engineering and Tr	ansport	
Prerequisites		
Knowledge: The student knows th	e basics of physics and	chemistry as well as the basics of
thermodynamics and fluid mechai	nics	

Skills: Student use of terminology terms in the field of mechanics, thermodynamics, physics and chemistry. Correct description of the observed phenomena, analysis of the obtained results and drawing conclusions.

Social competences. Work in an interdisciplinary team. Ability to lead a team and expand team knowledge.

Course objective

Getting to know the basic relationships showing the physical and chemical properties of gases.



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Course-related learning outcomes

Knowledge

Has knowledge in the field of physics, including the basics of classical mechanics, optics, electricity and magnetism, solid state physics, quantum and nuclear physics, necessary to understand specialist lectures in the field of the theory of construction materials and materials science, theory of machines and mechanisms, theory of electric drives and mechatronic systems.

Has basic knowledge in the field of chemistry, in the construction of the periodic table of elements and their properties, the theory of chemical bonds, organic and inorganic compounds, types of chemical reactions, chemical analysis: in the scope enabling understanding of lectures on metal and non-metal materials, protection sciences environment, fuels and lubricants, building materials and soil, biomechanics and biological materials processed by agricultural and food machinery.

Has basic knowledge of technical thermodynamics, ie the theory of thermodynamic changes, heat flow, thermal machines and heating, drying and cooling devices.

Skills

Can obtain information from literature, the Internet, databases and other sources. Can integrate the obtained information, interpret and draw conclusions from it, and create and justify opinions.

Can use learned mathematical theories to create and analyze simple mathematical models of machines and their elements, and simple technical systems.

Can develop a safety instruction for a simple and medium complex machine.

Has the ability to self-educate with the use of modern teaching tools, such as remote lectures, websites and databases, teaching programs, e-books.

Social competences

Is ready to critically assess his knowledge and received content.

Is willing to think and act in an entrepreneurial manner.

Is ready to fulfill professional roles responsibly, including:

- observing the rules of professional ethics and requiring this from others,

- caring for the achievements and traditions of the profession.

Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

The knowledge acquired during the lecture is verified on the basis of a written exam in the form of a test.

The skills acquired during the exercises are verified on the basis of a final test in the form of a written test.



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Programme content

Characteristics of the gas and liquid state. Thermodynamic properties: ideal, semi-perfect and real gas equations of state, compressibility factor. Viscosity of gases and liquids, depending on pressure and temperature. Influence of gases and liquids on pipeline materials. Influence of aggressive ingredients, anti-corrosion and anti-erosion protection. Combustion. Phase equilibria in multicomponent systems. Osmotic phenomena in two-component systems. Osmosis, dialysis. Donnan's membrane equilibria. Diffusion. Kinetics and mechanism of phase changes.

Teaching methods

Information and problematic lecture with a multimedia presentation. Exercises - solving problems.

Bibliography

Basic

1. H. Buchowski, W. Ufnalski "Fizykochemia gazów i cieczy", Wydawnictwa Naukowo -Techniczne, Warszawa 2012

- 2. J. Szargut: Termodynamika techniczna, PWN 1991
- 3.J. Molenda: Gaz ziemny, PWN 1999

Additional

1. K. Pigoń, Z. Ruziewicz: Chemia fizyczna, PWN 2012

Breakdown of average student's workload

	Hours	ECTS
Total workload	75	3,0
Classes requiring direct contact with the teacher	27	1,0
Student's own work (literature studies, preparation for	48	2,0
laboratory classes/tutorials, preparation for tests/exam, project		
preparation) ¹		

¹ delete or add other activities as appropriate